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device used to measure calorimetry. enthalpy. the heat content of a system at constant pressure. heat equals enthalpy change.  $q = \text{triangle } H$ . heat equals mass times change in heat times specific heat.  $q = m \cdot c \cdot \Delta T$ . thermochemical equation. chemical equation that includes enthalpy change.

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### SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

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