

# Pogil Phase Changes Answers

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...

Using the phase diagram of the liquid above, describe any changes in phase present when 1-120 is: 12. kept at 0°C while the pressure is increased from 1 atm to 5 atm (vertical line) 60b —S 13. Kept at 1.00 atm while the temperature is increased from 0°C to 125°C. (horizontal line) Created Date.

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Phase Changes Model 2: Temperature of a Substance as Heat is Added Over Time F POGIL 2005, 2006 3/4 Gas boiling D E Liquid B melting C

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Phase Changes Why? Most substances go through a phase change when heated or cooled. Molecules of a substance are held together in either the solid, liquid, or gaseous phase by intermolecular forces. It is necessary to discuss what is occurring at the molecular level in order to explain how an ice cube is melted or how water is boiled.

## **Phase Changes**

Create a cooling curve when given data and identify phase changes, E changes. Model 1 represents molecules in the three phases of matter. Based on what you already know about these phases, complete the table that follows. A word bank has been

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provided for the last column. D. E. G. F. MODEL 1: PHASE CHANGE PARTICLE DIAGRAM. Word Bank for Column ...

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Most substances will eventually go through a phase change when heated or cooled (sometimes they chemically react instead). Molecules of a substance are held together in either the solid, liquid, or gaseous phase by particle attractions. Though we are very familiar with seeing things like snow melting or water boiling, it is necessary to investigate what is occurring at the particle level in order to truly be able to explain how and why snow melts or water boils.

### **Chemistry Crunch #4.1 KEY Phase Changes**

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Phase Changes 6. On the molecular level, why is energy added in order to complete an endothermic phase change? (Refer to both Models in your answer.) marc 7. On the molecular level, why is energy released to complete an exothermic phase change? (Refer to both Models in your answer.) to 510uj down, so fly R-ud to 8. Comparing segments

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Phase Changes Why? Most substances go through phase change when heated or cooled. Molecules of a substance are held together in either the solid, liquid, or gaseous phase by intermolecular forces. It is necessary to discuss what is occurring at the molecular level in order to explain how an ice cube is melted or how water is boiled. Learning Objectives

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POGIL differs from other approaches in two particular ways. The

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first is the explicit and conscious emphasis on developing essential and purposeful process skills. The second is the use and design of distinctive classroom materials. Three defining characteristics of these materials are:

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Phase changes that fit this description include: Fusion, vaporization and sublimation. For any phase change that must require a cooling that allows molecules to move together another and strengthen intermolecular forces, potential energy decreases while kinetic energy remains constant.

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Phase Change POGIL Why? Most substances go through a phase change when heated or cooled. Molecules of a substance are held together in either the solid, liquid, or gaseous phase by intermolecular forces. It is necessary to discuss what is occurring at the molecular level in order to explain how an ice cube is melted or how water is boiled.

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phase change(s) would occur if I increased the temperature to 600Q C? At what temperature(s) would they occur? (NOTE: multiple answers needed for this question) 22) If I had a quantity o his substance at a pressure of 2.00 atm and a temperature of -1500 C, what phase change(s) would occur if I decreased the pressure to 0.25 atm? At what pressure(s)

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Population Growth Pogil. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Grace\_Faulkner45. Key Concepts: Terms in this set (34) ... During what phase of the growth curve in each diagram is the populations just beginning to colonize an area? Lag phase.

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Phase Change Reading Answer Key. Lets Get Physical With Phase  
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Burning Your Money! 15 minutes. Safety, Don't Take it Lightly. You and any student involved in this demonstration must wear goggles.

### **Phase Change Reading Answer Key - BetterLesson**

Using a graph, teach how the addition of heat causes ice to heat up and change from a solid to a liquid, and then to a gas. The graph shows how heating the ice causes the temperature to rise and how phase changes relate to this. Definitions of freezing point, melting point, boiling point, evaporatio...

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