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Note Taking Guide: Episode 701 Name:  
Kelly R. Lab results: 1 doz grains of rice  
= 0.3 g (Use this fact as a conversion  
factor.)? grains of rice = 1.94 g  
Avogadro's Number - the mole = the  
number  $6.022 \times 10^{23}$  Molar Mass the

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mass of one mole of any pure substance  
molar mass of an element (g/mol) equals  
the atomic mass of the element in  
grams molar mass of a compound  
equals the sum of the molar masses of  
the elements making up the compound .  
example problems: Find the molar  
masses of ...

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## **Note Taking Guide - Note Taking Guide Episode 701 Name ...**

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Matter © 2004, GPB 7.1 Lab results: 1  
doz grains of rice = \_\_\_\_\_g (Use this fact  
as a conversion

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## **Name**

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## **Liberty Union High School District / Overview**

CHEMISTRY: A Study of Matter © 2004,  
GPB 7.1 Note Taking Guide: Episode  
701 Name Lab results: 1 doz grains of  
rice = g (Use this fact as a conversion



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factor.) ? grains of rice = 1.94 g

Avogadro's Number - the mole = the number  $6.023 \times 10^{23}$  Molar Mass the mass of one mole of any pure substance molar mass of an element (g/mol) equals the atomic mass of the element in grams molar mass of a compound equals the sum of the molar mass of the element making up the compound

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example problems ...

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doz grains of rice = \_\_\_\_\_g (Use this fact  
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Season 1 Episode 701 | 25m 58s

Introduction to the Mole and Molar Mass:

The terms mole and molar mass are introduced and used in calculations.

Aired: 01/07/02

**Chemistry & Physics | Chemistry**  
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1 EP 701-Key.doc Author: Brent White  
Created Date: 6/23/2005 3:06:34 PM

**How many formula units are in 32.6**

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### **grams of potassium oxide?**

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