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11. fourth 12. transition elements 13. 32 14. valence electrons
15. electron affinity 16. electronegativity 17. ionization energy
18. 3 s² 3p⁴ 19. atomic radius 20 ...

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___ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ___ 6.

Mendeleev did not always list elements in his periodic table in

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order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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166–169 Section 5.1 Mastering Concepts 34. Define the following
terms. a. frequency Frequency is the number of waves that pass
a given point per second. b. wavelength Wavelength is the
shortest distance between

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CHAPTER. Chapter 5 Assessment pages 166169. Section 5.1
Mastering Concepts 34. Define the following terms. a. frequency
Frequency is the number of waves that pass a given point per
second.. b. wavelength Wavelength is the shortest distance
between equivalent points on a continuous wave.. c. quantum A

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quantum is the minimum amount of energy that can be lost or gained by an atom.

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5. Write the noble-gas notation for iodine. 6. Identify each atom.
a. $1s^2 2s^2 2p^1$ b. $[\text{Ar}]4s^1$ carbon sulfur 7. Write electron-dot structures for the following atoms. Chapter Assessment a. neon
b. hydrogen 28 Chemistry: Matter and Change Chapter 5 c. d.

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_____ 1.

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Chapter 5 - Electrons in Atoms - 5 Assessment - Page 156: 106 Answer The atomic mass of chlorine is very far from a whole because a weighted average of atomic masses of all of its isotopes is computed in determining its atomic mass.

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