

Chapter 5 Test The Periodic Law

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Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d.

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2. The principle that states that the physical and chemical properties of the elements are periodic functions of their atomic numbers is known as (a) the periodic table; (b) the periodic law; (c) the law of properties; (d) Mendeleev's law. 3. The elements with atomic numbers from 58 to 71 in the Periodic Table are referred to as

The Periodic Law ChapterTest

Chapter 5: The Periodic Table (Test Answers) Flashcards... The order of elements in the periodic table is based on. atomic mass. the number of protons in the nucleus. the number of neutrons in the nucleus. the electric charge of the nucleus. the number of protons in the nucleus. Semiconductors are elements that.

Chapter 5 Test The Periodic Table Multiple Choice Answer Key

chemistry chapter 5- the periodic table. periodic table. the synthetic elemnts are past what elem.... Dimitri Mendeleev. Moseley's contribution to the periodic.... listing of the known elements by property. uranium. father of the periodic table. arranged the elements by their number of protons rather than t....

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Chapter 5 TEST: The Periodic Table. Multiple Choice. Identify the choice that best completes the statement or answers the question. ____ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c.

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CHAPTER 5 REVIEW. The Periodic Law. SECTION 1. SHORT ANSWERAnswer the following questions in the space provided. 1. cIn the modern periodic table, elements are ordered. (a)according to decreasing atomic mass. (b)according to Mendeleev's original design. (c)according to increasing atomic number.

5 The Periodic Law

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____ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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