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mass to mass calculations . Limiting reactant limiting reactant determines the amount of product there is.

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Ms\_Li\_ClearView. Terms in this set (11) Stoichiometry. The calculation of quantities in chemical reactions. Is the number of atoms conserved in a balanced chemical equation? Yes  $N_2 + 3H_2 \rightarrow 2NH_3$

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Chemistry Chapter 12 Stoichiometry. stoichiometry. mole ratio. limiting reactant. excess reactant. the study of quantitative relationships between the amounts of.... in a balanced equation, the ratio between the number of moles.... a reactant that is



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totally consumed during a chemical reaction....

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An expanded version of the flowchart for stoichiometric calculations illustrated in Figure 11.4.1 is shown in Figure 12.2.1. We can use the balanced chemical equation for the reaction and either the masses of solid reactants and products or the volumes of solutions of reactants and products to determine the amounts

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of other species, as ...

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Calculations and Chemical Equations 139 The section ends with a summary of equation stoichiometry problems and shows how the skills developed in Section 10.1 can be mixed with the new skills developed in this section. Section 13.3 completes our process of describing equation

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