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following pairs of formulae is true for butanoic acid?

- Molecular Empirical  
Formula (A)  
 $C_4H_8O_2$   $CH_2O$  (B)  
 $C_4H_6O_2$   $CH_2O$  (C)  
 $C_4H_8O_2$   $C_2H_4O$  (D)  $C_4H_8O$   $C_2H_4O$

Combustion analysis of 0.18 g of an organic compound produces 0.396 g of carbon dioxide and 0.216 g of water.

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